

Silver Content Determination by Volhard Titration

The silver content is determined by precipitation titration with potassium thiocyanate KSCN as a titrant. The titration is monitored by a combined silver ring electrode.

Sample	Approx. 0.1 M silver nitrate sol., 3 mL AgNO ₃
Compound	Silver, Ag ⁺ M = 107.88 g/mol, z = 1
Chemicals	10 mL 0.1 M Nitric acid, HNO ₃ 50 mL Deionized water
Titrant	Potassium thiocyanate, KSCN c(KSCN) = 0.1 mol/L
Standard	5 mL 0.1 M AgNO ₃
Indication	DMi148-SC combined silver ring electrode
Chemistry	Ag ⁺ + SCN ⁻ → AgSCN
Calculation	R = Q·C/m, mol/L C = 1/z
Waste disposal	Filtrate the solution and dispose the precipitate as special waste.
Author, Version	Thomas Hitz, MSG AnaChem, Version 1.0

Preparation and Procedures

CAUTION

- Use safety goggles, a lab coat and wear gloves. If possible, work in a fume hood.
- Ensure accurate cleaning of sensor is sufficient after each titration.

Preparation of 0.1 mol/L KSCN:

- Weigh 2.430 g of solid KSCN and dissolve it in 100 mL deionized water in a 250 mL volumetric flask. After complete dissolution, fill up to the mark with deionized water.

Sample titration:

- Pipette 5 mL 0.1 mol/L into the titration beaker
- Add 10 mL 0.1 HNO₃
- Add 50 mL deionized water
- Connect the beaker to the titration stand, and start the method.
- After titration, rinse accurately with deionized water the electrode, the stirrer and the dispensing tube with water, and if necessary clean them with e.g. a paper tissue before starting next sample.

Remarks

- The application method has been developed for the mentioned sample. It may be necessary to optimize the method for your sample.
- The AgSCN-precipitate may be sticking onto the electrode, tube and stirrer surfaces. To improve the efficacy of the cleaning procedure, the precipitate can be dissolved with diluted ammonia solution.

Instruments

- Titration Excellence T50/T70/T90
(Other Titrators: depending on instrument type, manual operation and method changes are necessary)
- XS205 Balance

Accessories

- 1 x 10 mL DV1010 glass burette
- PP Titration beakers ME-101974
- LabX[®] pro titration
- Spatula

Results**0.1 M AgNO₃****All results**

Method-ID	AgDM148	
Sample	Silver	(1/3)
R1 (Content)	0.089	mol/L
Sample	Silver	(2/3)
R1 (Content)	0.090	mol/L
Sample	Silver	(3/3)
R1 (Content)	0.090	mol/L

Statistics

Method-ID	AgDM148	
R1	Content	
Samples	3	
Mean	0.090	mol/L
s	0.001	mol/l
srel	0.644	%

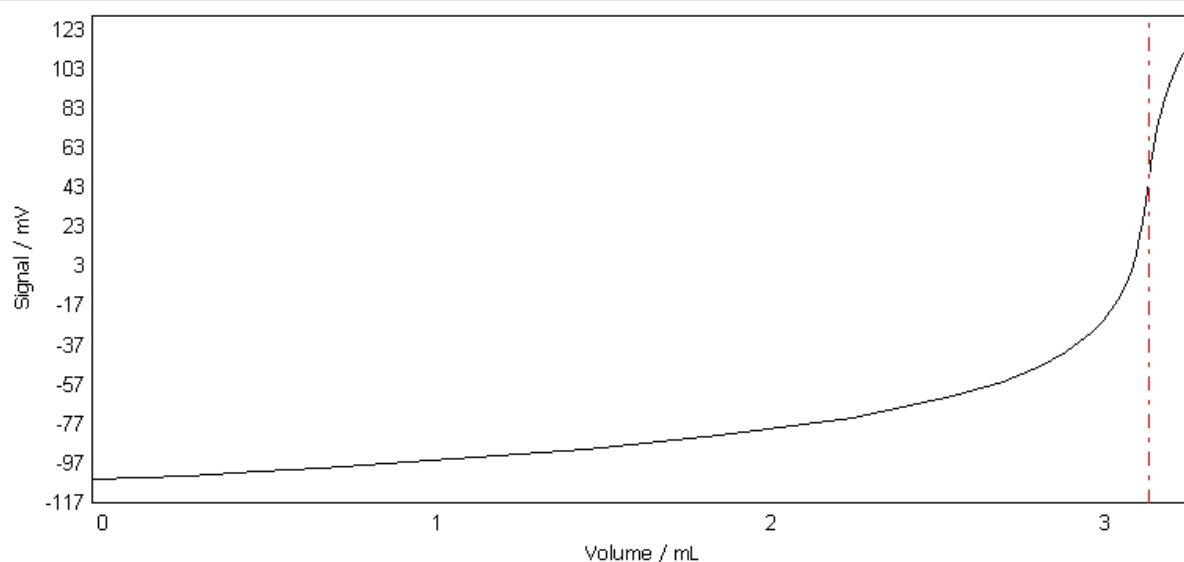
Titration curve

Table of measured values

	Volume mL	Increment mL	Signal mV	Change mV	1st deriv. mV/mL	Time s
EQP1	0.0000	NaN	105.3	NaN	NaN	0
	0.2855	0.2855	103.7	1.6	NaN	2
	0.4285	0.1430	101.6	2.1	NaN	5
	0.5000	0.0715	100.8	0.8	NaN	8
	0.6790	0.1790	99.1	1.7	NaN	16
	1.0790	0.4000	95.0	4.1	10.62	20
	1.4790	0.4000	89.9	5.1	12.66	22
	1.8790	0.4000	82.9	7.0	19.38	26
	2.2790	0.4000	73.7	9.2	34.35	29
	2.5780	0.2990	63.1	10.6	55.50	33
	2.7265	0.1485	55.7	7.4	76.38	36
	2.8390	0.1125	47.6	8.1	102.90	39
	2.9175	0.0785	40.6	7.0	135.97	42
	2.9885	0.0710	30.9	9.7	189.18	46
	3.0275	0.0390	24.8	6.1	245.61	49
	3.0715	0.0440	14.0	10.8	353.37	53
	3.0920	0.0205	7.4	6.6	450.07	56
	3.1115	0.0195	0.6	8.0	583.66	60
	3.1265	0.0150	8.8	8.2	715.51	63
	3.1375	0.0110	18.5	9.7	823.83	67
	3.1455	0.0080	25.5	7.0	950.42	70
	3.1550	0.0095	36.1	10.6	1077.87	74
	3.162279	NaN	44.8	NaN	1132.68	NaN
	3.1630	0.0080	45.7	9.6	1132.59	78
	3.1710	0.0080	55.1	9.4	1084.79	82
	3.1790	0.0080	63.1	8.0	980.99	85
	3.1885	0.0095	72.4	9.3	850.62	89
	3.1970	0.0085	79.2	6.8	751.00	92
	3.2095	0.0125	87.8	8.6	NaN	95
	3.2230	0.0135	95.0	7.2	NaN	99
	3.2420	0.0190	103.0	8.0	NaN	102
	3.2660	0.0240	111.1	8.1	NaN	105
	3.2955	0.0295	119.0	7.9	NaN	108

Comments

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Method

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001 Title
    Type                General titration
    Compatible with      T50 / T70 / T90
    ID                   AgDM148
    Title                AgCont
    . . .

002 Sample
    Number of IDs        1
    ID 1                 SILVER
    Entry type           Fixed volume
    Volume               3.0 mL
    Density              1.0 g/mL
    Correction factor    1.0
    Temperature          25.0°C

003 Titration stand (Manual stand)
    Type                Manual stand
    Titration stand     Manual stand 1

004 Stir
    Speed               35%
    Duration            15 s
    Condition           No

005 Titration (EQP) [1]
    Titrant
        Titrant          KSCN
        Concentration     0.1 mol/L
    Sensor
        Type             mV
        Sensor            DM148
        Unit              mV
    Temperature acquisition
        Temperature measurement No
    Stir
        Speed            35%
    Predispende
        Mode              Volume
        Volume            0.5 mL
        Wait time         5 s
    Control
        Control           User
        Titrant addition  Dynamic
        dE (set value)    8.0 mV
        dV (min)          0.008 mL
        dV (max)          0.4 mL
        Meas. val. Acquisition Equilibrium controlled
        dE                1.0 mV
        dt                1 s
        t (min)           3 s
        t (max)           30 s
    Evaluation and recognition
        Procedure         Standard
        Threshold         200.0 mV/mL
        Tendency          Positive
        Ranges            0
        Add. EQP criteria No
    Termination
        At Vmax           10.0 mL
        At potential      No
        At slope          No
        After number of recognized EQPs Yes
        Number of EQPs    1
        Combined termination criteria No

006 Calculation R1
    Result               Content
    Result unit          mol/L
    Formula              R1=Q*C/m
    Constant             C=1
    M                    M[None]
    z                    z[None]
    Decimal places       3
    Result limits        No
    Record statistics     Yes
    Extra statistical func. No
    Send to buffer        No
    Condition            No

007 End of sample

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