METTLER TOLEDO M113

## Sodium Determination with Ion Selective Electrode

Method for the sodium determination by direct measurement with a sodium ion selective electrode.

Sample	50 mL Sample solution	Preparation and Procedures			
	with 0.1 to 0.001 mol/L sodium	Milk: Dilute sample 1:10. To 50 mL of this add 1 mL ISA solution and measure.  Apple juice: To 50 mL undiluted sample add 1 mL ISA sol. and			
Substance	Sodium M = 22.99	1.5 mL $\rm NH_4OH$ 5% . The pH of the sample is adjusted to 7.5 with excess $\rm NH_4OH$ (For correct measurement with Na-ISE, the pH must be over 7). Wine, grape juice: To 50 mL undiluted sample add 1 mL ISA sol. and 1.5 mL $\rm NH_4OH$ 5% . The pH of the sample is adjusted to 8 with excess $\rm NH_4OH$ (the pH must be over 7). Mineral water: To 50 mL sample add 1 mL ISA solution. Drinking water: To 50 mL sample add 1 mL ISA solution and			
Chemicals	1 mL ISA solution (4 m NH <sub>4</sub> Cl and 1 m NH <sub>4</sub> OH)				
Titrant		measure. Due to the low sodium content, the standards 10-3 and 10-4 mol/L were used for calibration for this sample. <b>Bouillon:</b> Place a bouillon cube in 300 mL deion.water and warm till it has disolved. Let cool and use deion.water to adjust volume to 500 mL. Dilute this solution 1:20 resp. 1:50 with deion.water. To 50 mL add 1 mL ISA sol. and measure.			
Standard		Herb patè: Mix entire contents of can (125g) with 700 mL			
Instruments	DL67, DL70ES, DL77 Printer (HP Desk Jet 500)	deion. water for about 5 minutes. Decant fat. Fill to 1000 mL with deion. water. To 50 mL of this add 1 mL ISA solution and measure.  Infusion solution: Dilute sample 1:100 and to 50 mL of this add 1 mL ISA solution and measure.			
Accessories	Titration beaker ME101974 DT120 (Temp. sensor Pt100)				
Indication	DV222 andium inn colontive electrode	Remarks			
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		Reproducibility and Recovery Aqueous Na solutions made from NaCl were used (titrimetric standard). The Na-ISE was recalibrated daily with standards 0.1 and 0.001 g Na*/L. It was stored in a concentrated sodium solution. To 50 mL sample add 1 mL ISA sol. then measure. The solutions were produced daily.  Conc. Recovery RSD of sev. series(n=6) mol/L % Method A Method B 0.1 98-102% 0.2-0.5% 0.2-0.6%			
Chemistry		Reproducibility and Recovery Aqueous Na solutions made from NaCl were used (titrimetric standard). The Na-ISE was recalibrated daily with standards 0.1 and 0.001 g Na*/L. It was stored in a concentrated sodium solution. To 50 mL sample add 1 mL ISA sol. then measure. The solutions were produced daily.  Conc. Recovery RSD of sev. series(n=6) mol/L % Method A Method B 0.1 98-102% 0.2-0.5% 0.2-0.6% 0.01 95-100% 0.2-0.6% 0.3-0.7% 0.001 97-101% 0.2-0.8% 0.2-0.9% 0.0005 101-106% 0.2-0.4% 0.2-0.9% 0.0001 131-129% 0.2-0.9% 0.2-0.9% Result The differences between A and B are not decisive. Below 0.001 mol/L longer stir times (15-20 min) were necessary to get a good reproducibility. The concentration 0.001 mg/L leaves the linear range (see cal. curve). Thus, the useful concentration range was limited from 0.1 mol/L to 0.001 mol/L.			

## Results

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Sample	Dilution	Direct me	asurement with sodium		Content
		n	Mittelwert	RSD	(Producer spec.)
,	1:10 <b>am,</b> pasteuri	6 zed	400.8 mg/L	0.20%	430-490 mg/L*
	am, passouri	.200			
Milk, full cre	1:10 <b>am,</b> raw	6	433.8 mg/L	0.20%	430-490 mg/L*
Apple Juice	none	4	8.2 mg/L	0.64%	[30 mg/L*
Grape Juice, re	none <b>d</b>	5	23.8 mg/L	0.66%	[30 mg/L*
Wine, red Italian	none	5	12.2 mg/L	0.63%	10-80 mg/L*
Wine, White, S	none wiss	5	8.4 mg/L	2.0%	10-80 mg/L*
Mineral Water	none	6	8.17 mg/L	0.22%	8.2 mg/L
Drinking Water, S	none chwerzenbach	5	3.81 mg/L	0.35%	
Bouillon sodium re		5	468.5 mg/L	0.35%	[1.2 g/L
Bouillon normal	1:50	5	4110 mg/L	0.41%	-4 g/L
Herb pate		6	175.9 mg/kg	0.52%	[0.4 g/kg
Infusion Solution		5	142.5 mmol/L	0.72%	141.5 mmol/L

<sup>\*</sup> Obtained from: Schweizerisches Lebensmittelbuch

Method A stirs for 5 minutes then takes 6 measurements with an interval of at least 10 seconds. The sodium concentration is calculated from the mean of these 6 measured values.

Method B measures once after a stir time of 5 minutes. The sodium concentration is calculated from this. If the sample is diluted for the measurement, the dilution factor can ve entered as a correction factor (f) for each sample. This will be used in the calculation.

Table of measured values	Titration curve

## Method Method A Calculation (Direct measurement with mean value) Method ID . . . . . . . . . . . . Na0A Calculation Sample Number samples . . . . . . . . . . . . . . . . . . Number samples 6 Titration stand Stand 1 Entry type Fixed volume U Volume [mL] 50.0 ID1 ... Molar mass M 0.0 Equivalent number z 1 Temperature sensor TEMP A Statistics Stir Outlier test . . . . . . . . Yes Speed [%] . . . . . . . . . . . . 50 Time [s] . . . . . . . . . . . . 600 Statistics easure . . . . . . . . . . . . . . Na+ -sensor Sensor . As installed ecord Output unit . . . . . . . . . . . Printer All results . . . . . . . . . . Yes Sensor . . . . . . . . . . . . . . . Na+ -sensor (Simple direct measurement) leasure Sensor . . . . . . . . . . . . . Na+ -sensor Sample mple 6 Number samples 6 Titration stand ST20 Entry type Fixed volume U Volume [mL] 50.0 ID1 ... Molar mass M 22.99 Equivalent number z 1 Temperature sensor TEMP A ir Sensor Na+ -sensor Unit of meas As installed dE [mV] 0.2 dt [s] 5.0 t(min) mode Fix t (min) [s] 10.0 t (max) [s] 90.0 Measure Sensor . . . . . . . . . Na+ -sensor Sensor Na+ -sensor Unit of meas As installed dE [mV] 0.2 dt [s] 5.0 t(min) mode Fix t (min) [s] 10.0 t (max) [s] 90.0 Speed [%] Measure Measure Calculation Result name R=E Formula R=E Constant p(Na+) Decimal places 3 Measure Sensor Calculation Calculation Statistics lculation Result name . . . . . . . . . . Na+ -single Outlier test . . . . . . . . . . Yes ecord Output unit . . . . . . . . . . . . Printer All results . . . . . . . . . . . Yes